
As you've probably noticed at some point or another, molecules are all around us. This includes the air we breathe, the food we eat, and even the clothes that cover our bodies. Molecules are made up of three or more atoms that typically have different elements in them. These can be categorized into two groups: molecular compounds and ionic compounds. Molecular compounds are held together by the covalent bonds that we talked about previously. Ionic compounds, on the other hand, are held together by a different type of bond known as a metallic bond. Metallic bonds are formed when a metal atom loses one or more electrons in combination with a nonmetal atom gaining them until all have an equal charge. This bond is incredibly strong and allows for ionic compounds to have melting and boiling points much higher than the corresponding molecular ones. In order to help you remember these distinctions between molecular and ionic compounds, let's review some definitions. Molecular Compounds Ionic Compounds Now that you're aware of the difference between molecular and ionic compounds, you can compare them.

Electron Configuration Gizmo Answer Key (K-H) (from the back) As you've probably noticed at some point or another, molecules are all around us. This includes the air we breathe, the food we eat, and even the clothes that cover our bodies. Molecules are made up of three or more atoms that typically have different elements in them. These can be categorized into two groups: molecular compounds and ionic compounds. Molecular compounds are held together by the covalent bonds that we talked about previously. Ionic compounds, on the other hand, are held together by a different type of bond known as a metallic bond. Metallic bonds are formed when a metal atom loses one or more electrons in combination with a nonmetal atom gaining them until all have an equal charge. This bond is incredibly strong and allows for ionic compounds to have melting and boiling points much higher than the corresponding molecular ones. In order to help you remember these distinctions between molecular and ionic compounds, let's review some definitions. Molecular Compounds - compounds composed of molecules made up of covalently bonded atoms Ionic Compounds - compounds composed of ions held together by metallic bonds Now that you're aware of the difference between molecular and ionic compounds, you can compare them.

And this is the answer key for the electron configurations I hope it helped!

The chart below contains electron configuration data for various elements. Along with each configuration is a number that tells how many electrons are found in the subshell that contains that electron configuration. You should be able to find this number on your subshell chart.

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